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**SAMPLE ANSWER:
Organisms use carbon
compounds to form
four types of
molecules: lipids,**

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carbohydrates, nucleic acids, and proteins.

SAMPLE ANSWER: A lot of what that happens in an organism is based on chemical reactions.

SAMPLE ANSWER: Enzymes are proteins that speed up chemical reactions that take place in cells.

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College Students Lack Scientific Literacy, Study Finds
Specifically, this research indicates that students do not understand the processes that transform carbon compounds nor do they understand that most of a tree's growth comes from atmospheric CO₂ and water. Lab 1 is designed to address these misconceptions

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and lack of
understanding of basic
carbon cycle processes
that underpin the
carbon ...

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**Answer to: Determine
the oxidation number
of carbon in the
compound $\text{Na}_2\text{C}_2\text{O}_4$.
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**Answer: For Carbon, %
of Carbon = $n \times$
molar mass of element
molar mass of
compound $\times 100$ % of
Carbon = $1 \times 12.011 \text{ g /}$
 $\text{mol} / 44.009 \text{ g / mol} \times$
 100 % of Carbon = 27.3
% For Oxygen % of
Oxygen = $2 \times 15.999 \text{ g /}$
 $\text{mol} / 44.009 \text{ g / mol} \times$**

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**100 % of Oxygen =
72.7 % If we can
compute the percent
composition by mass
and the molar mass of
the compound from the
...**

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**Carbon is a crucial
element of our world.
Everything that
surrounds is probably
a compound of carbon.**

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Foods, fuels, textiles, drugs everything is either a compound of carbon or its substituent. So let us study about some important carbon compounds.

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In naming compounds, the first step is to identify the longest continuous carbon

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chain (except for ring systems). Next, identify the main functional group if any. With the known number the carbon...

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Measuring Matter 1.
pair 2. 5 3. dozen 4.
gross 5. 200 6. ream 7.
6,000,000,000 8. 0.5
mol 9. 6.02 10²³ 10.
four moles 11. 6.02 10²³
Cu atoms 12. 1 mol Cu**

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12. 4 23 4 1 mol CH

6.02 10 molecules CH

13. 23 1 mol Xe

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Carbon compounds are chemical substances that contain carbon atoms bonded to any other element. There are more carbon compounds than for any other element except hydrogen. The

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majority of these molecules are organic carbon compounds (e.g., benzene, sucrose), although a large number of inorganic carbon compounds also exist (e.g., carbon dioxide).

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**carbon - C nitrogen- N
oxygen - O fluorine - F
phosphorus - P sulfur -**

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S chlorine - Cl
selenium - Se
bromine - Br
iodine - I
Names and formulas of some binary covalent compounds.

Name	Formula
water	H ₂ O
methane	CH ₄
ammonia	NH ₃
ethane	C ₂ H ₆
propane	C ₃ H ₈

24
Study Guide for An Introduction to Chemistry Prefixes

Number of atoms	Prefix
1	mono-

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this chapter comes
from chapter 8.**

**General reading about
ATP and electron
transport chains comes
from chapter 6.**

Questions to think

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about... these questions are geared strictly toward preparing for your quiz. The other list you were provided has material that should help guide your reading and prepare more for the full AP exam.

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